

AUB
Physics Department

Phys 210
Final exam

Jan. 23, 2006
Time:100 min

Name : _____

ID. NO : _____

Major : _____

INFO:

- ◆ No make up of this exam
- ◆ All questions are obligatory. Only one answer can be crossed in case of multiple questions. Your answer must be identified uniquely.

<u>Page</u>	<u>Grade</u>	
3) _____	/ 22	4 questions
4) _____	/ 21	4 question
5) _____	/ 35	7 questions
6) _____	/ 12	Problem 1
7) _____	/ 10	problem2

Total:

Physical Constants

$$R=8.31 \text{ J/ (Mol .K)}$$

$$1 \text{ atm}=1.01 \times 10^5 \text{ N/m}^2$$

$$c_w = 4.19 \times 10^3 \text{ J/(Kg .K)}$$

$$L_{\text{ice}} = 3.33 \times 10^5 \text{ J/Kg}$$

$$L_{\text{steam}} = 2.26 \times 10^6 \text{ J/Kg}$$

$$\rho_{\text{steam}} = 0.598 \text{ kg/m}^3$$

$$\rho_{\text{water}} = 1000 \text{ Kg/m}^3$$

Gas constant

atmospheric pressure

specific heat of water

Latent heat of fusion

Latent heat of vaporization

density of steam at 100 °C

density of water

- ♥ (5 points) Five moles of an ideal gas undergo free expansion from an initial volume of 20 cm^3 to a final volume of 100 cm^3 . The change in entropy ΔS of this gas (in J / K) is:
- a) 191 b) 52 c) 67
 d) 67 e) 71

- ♥ (5 points) Three engines operates between reservoirs separated by 300 K.
 Engine A operates between 1000 K and 700 K.
 Engine B operates between 800 K and 500 K.
 Engine C operates between 600 K and 300 K.

- a) All these engines have the same efficiency , because the temperature difference is the same
 b) Engine A has the highest efficiency
 c) Engine C has the highest efficiency
 d) Engine B has the highest efficiency

- ♥ (6 points)

1) Which of the following is true for the entropy change (ΔS) of a system that undergoes a reversible adiabatic process?

- a) $\Delta S < 0$ b) $\Delta S > 0$ c) $\Delta S = 0$

2) What is the answer if the process is adiabatic but irreversible?

- a) $\Delta S < 0$ b) $\Delta S > 0$ c) $\Delta S = 0$

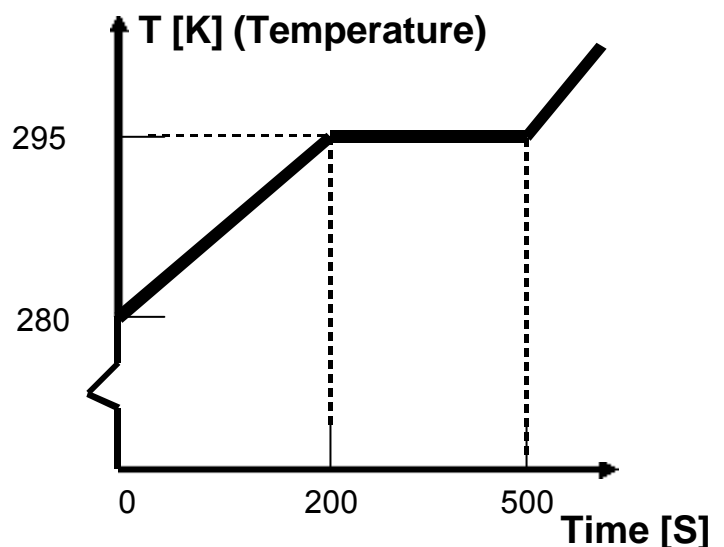
- ♥ (6 points) Heat is added to 0.25 Kg of a solid substance of yet unknown specific heat (C_s) at a steady (constant rate of **50 J/S** (Joules/second)). The temperature T of the substance which is initially solid is shown in the adjacent Figure as a function of time

1) (3 points) The specific heat of this solid is (in **J/(Kg .K)**):

- a) 130 b) 750
 c) 2700 d) 4000

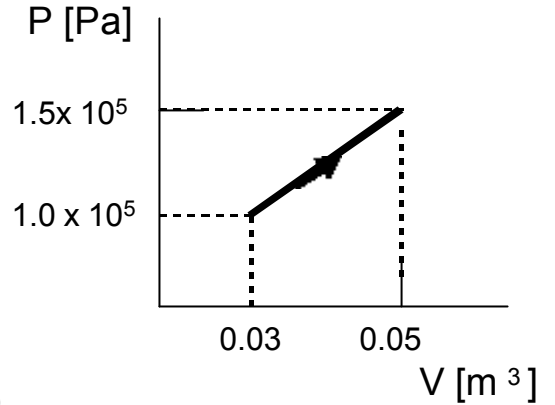
2) (3 points) The latent heat of fusion is (in **J/Kg**):

- a) 3.35×10^2 b) 6×10^4
 c) 4×10^4 d) 200



Score: 22

♥ (6 points) An **ideal monatomic gas** of 1.8 moles undergoes the process shown in the Figure



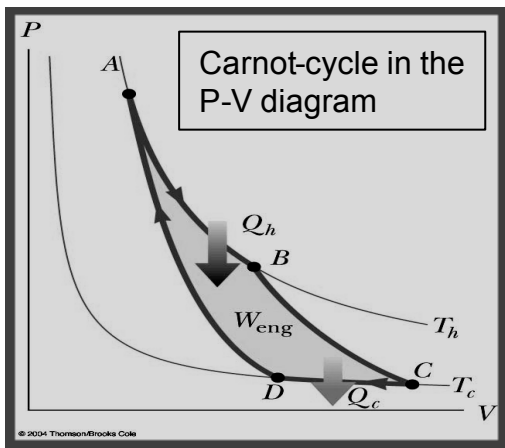
1) The temperature change is: (3 points)

- a) 200 → 300 K b) 200 → 333 K
 b) 300 → 500 K c) 200 → 500 K

2) The internal energy change is about (in J): (3 Points)

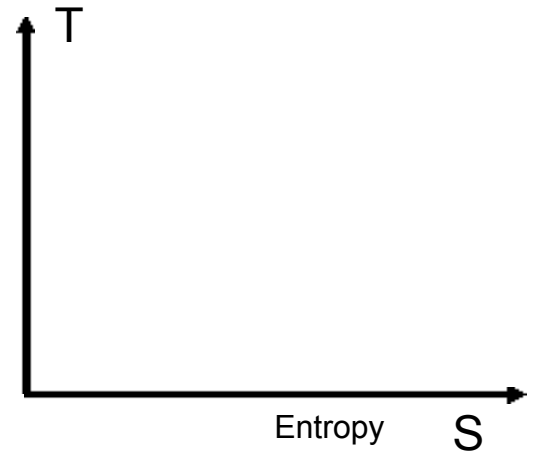
- a) 4900 b) 6731 c) 3030 d) 4487
 e) None of the above, my answer is: _____

♥ (4 points) A Carnot cycle operating as heat engine is shown in the adjacent Figure. It can be represented on a T-S diagram (T=Temperature, S=Entropy).



Question

Draw the Carnot Cycle in the T-S Diagram: start at A → B → C → D → A



♥ (6 points) An amount of 5.0 Kg of water at T=288 K is mixed with 3.0 Kg of water at T=348 K in an insulated container. What is the change of entropy (ΔS) of the system in (J/K)?

- a) 1.05×10^5 b) 138 c) 142 d) 3000
 e) none of the above, my answer is: _____

♥ (5 points) In a Carnot cycle, the entropy change is 2.57 (J/K) for the reversible isothermal expansion at $T_h = 500$ K. What is the change in entropy for the reversible isothermal compression at $T_c = 300$ K?

- a) +2.57 J/K b) 0.0 J/K c) -2.57 J/K d) 1.54 J/K e) +1.54 J/K

Score: 21

- ♥ (4 points) You see two helium balloons floating close to each other (distance between them is about 2.0 cm). The balloons are fixed by strings to a table. You blow through the small space between the balloons. What will happen to them?
- a) They move away from each other b) They are not affected at all
- c) They move toward each other
- ♥ (5 points) A beam of unpolarized light in air is incident at an angle of 58.6° (with respect to the normal) on a plane glass surface of unknown refractive index. The reflected beam is completely polarized. The refractive index of the glass is:
- a) 1.60 b) 1.64 c) 1.33 d) 1.50
- ♥ (4 points) Suppose a **Michelson interferometer** is adjusted such that the monochromatic light beam produces a maximum in intensity at the detector. If the movable mirror is moved a distance ΔL and during this motion the intensity decreases to minimum. What is the wavelength of the light?
- a) $\Delta L/4$ b) $\Delta L/2$ c) ΔL d) $2 \Delta L$ e) $4 \Delta L$
- ♥ (5 points) Monochromatic light from a He-Ne laser ($\lambda = 632.8 \text{ nm}$) is incident on a diffraction grating containing 5000 lines/cm. The angle of the first-order maximum is:
- a) 18.4° b) 39.2° c) 14.6° d) 27.7° e) 13.9°
- ♥ (5 points) A length of organ pipe is closed at one end. If the speed of sound is 344 m/s. What length of pipe (in cm) is needed to obtain a fundamental frequency of 50 Hz?
- a) 1.72 b) 86 c) 344 d) 172 e) 688
- ♥ (6 points) A string is stretched and fixed at both ends, 200 cm apart. If the density of the string is 0.015 g/cm, and its tension is 600 N, what is the wavelength (in cm) of the fundamental harmonic?
- a) 600 b) 400 c) 800 d) 200
- ♥ (6 points) A thin film of a refractive index $n_f = 1.29$ is to be placed on a glass plate of $n = 1.50$. The minimum thickness for the film such that the reflection of normally incident light with $\lambda = 600 \text{ nm}$ is minimized is ($1 \text{ nm} = 1 \text{ nanometer} = 10^{-9} \text{ m}$):
- a) 232 nm b) 58 nm c) 116 nm d) 465

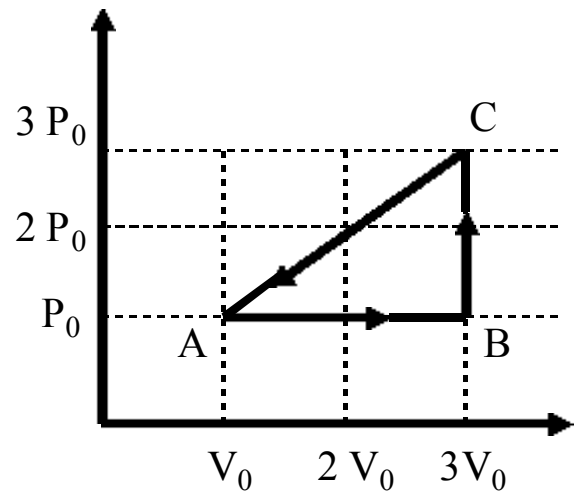
Score: 35

P1) (12 points)

A substance is taken through the Cycle shown in Figure.

Given:

P_0 [pa]	V_0 [m ³]	Q_{AB} [J]	Q_{BC} [J]
10^5	10^{-3}	450	200



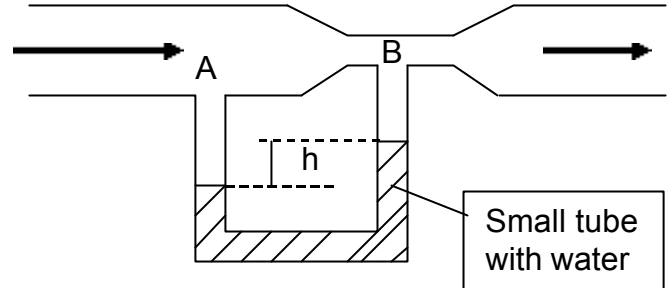
Q_{AB} = heat absorbed by the system from A → B,
 Q_{BC} = heat absorbed by the system from B → C

- a) (4 points) What is the work done on the gas during the cycle?
- b) (4 points) If the internal energy in the state B is $(E_{int})_B = 200$ J, what is the $(E_{int})_C$?
- c) (4 points) What is the amount of heat used during the process C → A.
Is this heat absorbed or rejected by the system?

P2) (10 points)

In the Figure, air moves the horizontal tube. At point A the air speed is $v_A = 2.0 \text{ m/s}$ and the radius is of the tube is $r_A = 0.05 \text{ m}$. At point B the radius is $r_B = 0.02 \text{ m}$. The small tube contains some water.

The density of the air is $\rho_a = 1.29 \text{ Kg/m}^3$.
Take gravitational acceleration $g = 10 \text{ m/s}^2$



a) Calculate the speed at point B, v_B .

b) Find the height h indicated in the Figure between the water level in the small tube