

3.15 In order to determine the APF for U, we need to compute both the unit cell volume ( $V_C$ ) which is just the product of the three unit cell parameters, as well as the total sphere volume ( $V_S$ ) which is just the product of the volume of a single sphere and the number of spheres in the unit cell ( $n$ ). The value of  $n$  may be calculated from Equation 3.5 as

$$n = \frac{\rho V_C N_A}{A_U}$$

$$= \frac{(19.05 \text{ g/cm}^3)(2.86)(5.87)(4.95)(\times 10^{-24} \text{ cm}^3)(6.023 \times 10^{23} \text{ atoms/mol})}{238.03 \text{ g/mol}}$$

$$= 4.01 \text{ atoms/unit cell}$$

Therefore

$$\text{APF} = \frac{V_S}{V_C} = \frac{(4)\left(\frac{4}{3}\pi R^3\right)}{(a)(b)(c)}$$

$$= \frac{(4)\left[\frac{4}{3}(\pi)(1.385 \times 10^8 \text{ cm})^3\right]}{(2.86)(5.87)(4.95)(\times 10^{-24} \text{ cm}^3)}$$

$$= 0.536$$